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properties of the gases (pressure, volume, temperature, or number of gas particles) inside and outside the balloon. Chapter 11 Gases - An Introduction to Chemistry CHAPTER 11 REVIEW Gases SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. b Pressure surf f a o c r e ce area. For a constant force, when the surface area is tripled the pressure is (a) doubled. (b) a third as much. (c) tripled. (d) unchanged. 2. d, c, a, b Rank the following pressures in increasing order. (a) 50 kPa (c) 76 torr (b) 2 atm (d) 100 N/m<sup>2</sup> 3. mc06se cFMsr i-vi - Ed W. Clark High School Start studying Chapter 11- Gases: Section 1: Gases and Pressure. Learn vocabulary, terms, and more with flashcards, games, and other study tools. Chapter 11- Gases: Section 1: Gases and Pressure ...this theory explains some of the properties of ideal gases. In this chapter, you will study the predictions of kinetic-molecular theory for gases in more detail. This includes the relationship among the temperature, pressure, volume, and amount of gas in a sample. SECTION 11.1 Key Terms pressure newton barometer millimeters of mercury SECTION 11.1 Gases and Pressure - Pickford High School CHAPTER 11 REVIEW Gases SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances: increase a. temperature increases, volume stays the same decrease b. volume increases, temperature stays the same mc06se cFMsr i-vi - Ed W. Clark High School Download chapter 11 review gases section 2 answers - Bing book pdf free download link or read online here in PDF. Read online chapter 11 review gases section 2 answers - Bing book pdf free download link book now. All books are in clear copy here, and all files are secure so don't worry about it. Chapter 11 Review Gases Section 2 Answers - Bing | pdf ... Chapter 11 Review Gases Section 2 Answers.pdf - search pdf books free download Free eBook and manual for Business, Education, Finance, Inspirational, Novel, Religion, Social, Sports, Science, Technology, Holiday, Medical, Daily new PDF ebooks documents ready for download, All PDF documents are Free, The biggest database for Free books and documents search with fast results better than any online ... Chapter 11 Review Gases Section 2 Answers.pdf | pdf Book ... Modern Chemistry 93 Gases CHAPTER 11 REVIEW Gases SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. \_\_\_\_ Pressure = . For a constant force, when the surface area is tripled the pressure is (a) doubled. (b) a third as much. (c) tripled. (d) unchanged. 2. \_\_\_\_ Rank the following pressures in increasing order. CHAPTER 11 REVIEW Gases - manasquanschools.org Download chapter 11 review gases section 2 answers - Bing book pdf free download link or read online here in PDF. Read online chapter 11 review gases section 2 answers - Bing book pdf free download link book now. All books are in clear copy here, and all files are secure so don't worry about it. Chapter 11 Review Gases Section 4 Answers Chapter 11 - Gases - An Introduction to Chemistry 182 Study Guide for An Introduction to

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tripled the surface area pressure is (a) doubled. as much. (c  
ripled. 7-0 (d) unchanged. Rank the following pressures in  
increasing order. (c) 76 torr (a) 50 kPa O, OOictbv-x  
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the properties of mixtures of gases. To describe calculations that  
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mixtures. This section describes how mixing gases affects the  
properties of the resulting mixture.

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tripled the pressure is (a) doubled. (b) a third as much. (c) tripled.  
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increasing order. (a) 50 kPa (c) 76 torr (b) 2 atm (d) 100 N/m<sup>2</sup> 3.

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Kinetic Theory of Gases Part 5 General Chemistry 1**

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Part 2 of 12 Intermolecular Forces Which gas equation do I use?  
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the following questions in the space provided. 1. State whether  
the pressure of a fixed mass of gas will increase, decrease, or  
stay the same in the following circumstances: increase a.  
temperature increases, volume stays the same decrease b.  
volume increases, temperature stays the same

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If a gas and a liquid are the same temperature and pressure, diffusion occurs much faster in the gas because. A. there are more elastic collisions between the particles in a gas. B. gases are more compressible. C. the particles move faster in a gas and there is a greater distance between them.

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inside and outside the balloon.

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#### *SECTION 11.1 Gases and Pressure - Pickford High School*

this theory explains some of the properties of ideal gases. In this chapter, you will study the predictions of kinetic-molecular theory for gases in more detail. This includes the relationship among the temperature, pressure, volume, and amount of gas in a sample.

SECTION 11.1 Key Terms pressure newton barometer millimeters of mercury

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